STRUCTURE OF CYCLONERODIOL

Shigeo Nozoe, Masami Goi, and Naoko Morisaki Institute of Applied Microbiology, University of Tokyo

Bunkyo-ku, Tokyo

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During the course of the investigations on the biosynthesis of fungal isoprenoids, a new sesquiterpenediol, cyclonerodiol has been isolated from a certain strain of <u>Trichothesium</u> species fungi. We here report the structure of cyclonerodiol possessing a novel type of carbon skeleton in the field of sesquiterpenoids.

Cyclonerodiol 1, C₁₅H₂₈O₂, mass peak at m/e 222 (M-18)¹⁾, $[\alpha]_{\rm D}$ - 20° (CHCl₃), rrt., 0.76²⁾, shows nmr signals³⁾ for five methyl groups at 0.97 (d, J=7 Hz), 1.08 (s), 1.18 (s) 1.59 (bs) and 1.65 (bs) indicating that this compound might possess an unusual structural feature as a sesquiterpenic compound. The nmr spectrum of 1 shows an olefinic proton signal at 5.03 (bt, J=7 Hz). In the mass spectrum of 1, fragmentation leading to the ions $\frac{4}{10}$ m/e (a) 113, 95 (b) 127, 109 (c) 157, 139, 121 and (d) 83 were observed. These fragmentation patterns indicate the electron impact cleavage of the C-C bond depicted in the following formula and the corresponding ions derived by loss of H₂O.



Heating <u>1</u> at 135° with phthalic anhydride gives a hydrocarbon, <u>2</u>, C_{15H24} (M⁺, 204), which absorbs three moles of hydrogen to afford a saturated compound, <u>2</u>, C_{15H30} (M⁺, 210). The second dehydration product <u>4</u>, C_{15H260} (M⁺, 222) shows new nmr signals at 1.67 (bs, 3H) and 5.14 (bt, 1H) due to a -C(CH₃)=CH- grouping. The prominent peaks in the mass spectrum of 4 are at m/e (a) 95, (b) 127, 109 (c) 139, 121 and (d) 83. Hydrogenation of 4 with two moles of hydrogen affords a saturated monoalcohol, 5, C15H300 (M⁺, 226), showing nnr signals of four secondary methyl groups at 0.86- 0.92 and a tertiary methyl at 1.07. The mass spectrum of 5 exhibits peaks at m/e (a) 97 (b) 129, 111 (c) 141, 123 and (d) 85, suggesting a ring must be in portion a. The main mass fragmentation of the third dehydration product 6 was very similar to that of 4, but four olefinic methyl signals occur at 1.60 and 1.67 in the nmr spectrum of $\underline{6}$.

The spectroscopic properties described above indicate that the Ceportion (<u>b</u>) must be a side chain possessing a tertiary methyl, a hydroxyl, and a double bond⁵⁾: the C7-portion (<u>a</u>) must contain a hydroxyl, two methyl substituents, and hence, a five membered ring.



Ozonolysis of $\underline{1}$ yields a mixture of the trisnor compounds $\underline{7}$, $\underline{8}$, and $\underline{9}$. The mixture of the products shows an aldehydic proton signal at 9.15, a hemiacetal proton signal at 5.38 in the nmr spectrum and an absorption band at 1766 cm⁻¹ in the ir spectrum. Treatment of the mixture with Jones' reagent

No.15

in acetone gives a single product, 9, m.p. 66-8°, $[\alpha]_{\rm D}$ -52.5° (CHOl3), C12H20O3 (M⁺, 212, intense peaks at m/e 99 and 95); nmr; 1.04 (d, J=7, 3H), 1.26 (s, 3H), and 1.39 (s, 3H), and 2.26 (m, 2H) exhibiting ir absorption at 1766 $\rm cm^{-1}$ (S-Dehydration of the lactone 9 with thionyl chloride in pyridine lactone). affords compound 10, C12H1802 (M⁺, 194); nmr, 1.62, 1.63 (two olefinic methyls), and 1.28 (methyl on a lactone ring); ir, 1766 cm⁻¹. This finding indicates that two methyl groups on the five membered ring are adjacent. Heating of the lactone 9 at 130° with phthalic anhydride gives the lactone 11, C12H18O2 (M⁺, 194); nmr, 1.62 (bs, 3H), 1.23 (d, J=7 Hz, 3H), and 5.13 (m, 1H); ir, 1766 cm⁻¹ along with the afore-mentioned lactone 10 in a ratio of The multiplicity of the signal of the trisubstituted olefinic proton 7:3. at 5.13 of lactone 11 reveals the presence of an allylic methylene group in 11. All of the experimental results mentioned support that cyclonerodiol possesses the planary structure 1.

As a minor constituent of the metabolite, compound 12, rrt, 1.20^{2} ; $C_{15H_{22}O_3^{(1)}}$, $[\alpha]_D^{(-)}-36.7^{\circ}$ (CHCl3), was also isolated from the same source. The nmr spectrum of 12 shows five methyl signals at 1.03 (d, J=7 Hz), 1.13 (s), 1.17 (s), 1.21 (s), 1.26 (s) and a methine proton at 3.74. The mass fragmentation leading to the ion m/e 197, 179, 143 (base peak), 113, 95 and 59 can be explained as shown below⁶. Oxidation of 12 with Jones' reagent⁷⁾ gives γ -lactone which was proved to be identical with 2.



1295

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- The mass spectrum did not show the molecular ion peak, but gave (M-18) ion peak ⁴. Satisfactory analytical data were obtained for this compound.
- 2. Herein rrt indicates relative retention time (on the 1.5% 07-17 on Shimalite W, 4mm x lm column, N2 flow rate, 40 ml/min., column temp. 120°) to trans, trans-farmesol(1.00, rt, 9.3 min.)
- 3. All nmr spectrum were measured in CCl4 solution and signals given in $\ensuremath{\S}\xspace$ -value.
- 4. The metastable peaks were observed at expected positions.
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